

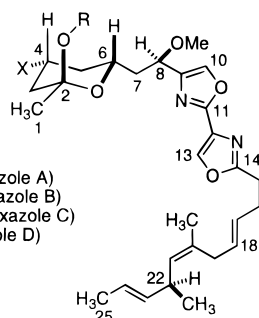
Total Synthesis of (–)-Hennoxazole A

David R. Williams,* Dawn A. Brooks, and Martin A. Berliner

Department of Chemistry, Indiana University
Bloomington, Indiana 47405

Received February 15, 1999

Since 1986, there has been a dramatic increase in the number of examples and structural complexity of bioactive natural products containing the oxazole ring. Marine organisms are rich sources of these novel metabolites.¹ Bisoxazoles, in which the two rings are directly linked by a single bond, are exemplified by the hennoxazoles (**1–4**), first isolated from the sponge *Polyfibrospongia* in 1991.² Hennoxazole A (**1**) displays potency against herpes simplex virus type 1 and peripheral analgesic activity comparable to that of indomethacin. The absolute configuration of **1** and the issue of relative stereochemistry at C₈ and C₂₂ have been resolved following synthesis of the (+)-enantiomer of **1** by Wipf and Lim.³ Herein we report an efficient strategy providing for an enantiocontrolled convergent total synthesis of (–)-hennoxazole A.

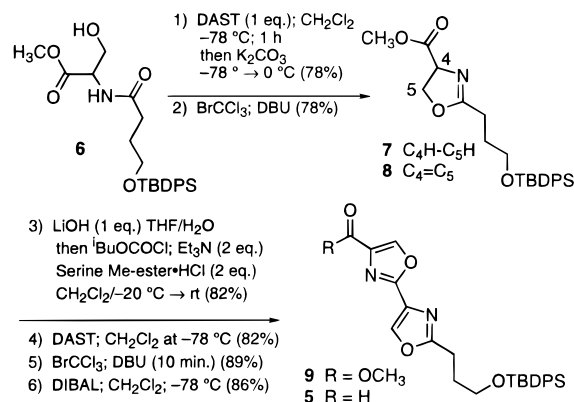


- 1 X = OH; R = CH₃ (Hennoxazole A)
2 X = OH; R = C₂H₅ (Hennoxazole B)
3 X = OH; R = n-butyl (Hennoxazole C)
4 X = H; R = CH₃ (Hennoxazole D)

Our plans sought the direct incorporation of a fully functionalized tetrahydropyran segment (C₁–C₇) with C-linkage to the heterocyclic core and creation of C₈ chirality. Concerns for stability of the nonconjugated triene, including the remote stereochemistry of the C₂₂ bis-allylic methine, suggested attachment of the C₁₈–C₂₅ portion in the final stages.

The 2,4-disubstituted bisoxazole **5** was assembled as summarized in Scheme 1. Using a mixed anhydride procedure, the coupling of 4-(*tert*-butyldiphenylsilyloxy)butanoic acid⁴ and (±)-serine methyl ester hydrochloride afforded **6**. Cyclization to oxazoline **7** occurred in a single step with diethylaminosulfur trifluoride (DAST)⁵ at –78 °C. Oxidation with bromotrichloromethane and DBU cleanly effected dehydrogenation to oxazole **8**.⁶ Reiteration of this protocol illustrates a general and highly effective synthesis of these heterocycles. Studies of the addition of reactive nucleophiles to aldehyde **5** led to competing ring deprotonation.⁷ However, the application of a mild asymmetric

Scheme 1



allylation strategy was developed to yield functionalized homoallylic alcohols⁸ based upon the pioneering efforts of E. J. Corey.⁹ Adaptation of this concept has expeditiously led to construction of the C₁–C₁₇ portion of hennoxazole A as shown in Scheme 2.

Formation of the (*R*)-homoallylic C₈ alcohol **14** was achieved by transmetalation of optically pure stannane **12** with (*R,R*)-bromoborane **13**⁹ via allylic transposition to yield an intermediate borane for facile condensation with aldehyde **5**. Stereocontrol is induced from the 1,2-diphenylethane sulfonamide auxiliary (10.5:1 dr), and is predicted from a chairlike transition state with minimized steric repulsions.¹⁰ Stannane **12** was conveniently prepared via copper-catalyzed Grignard addition starting from 2-bromo-3-trimethylsilylpropene¹¹ and nonracemic epoxide **10**.¹² The superior reactivity characteristics of allylstannane **12** were required as the silane **11** failed to undergo transmetalation with **13**, and direct attempts for Lewis acid mediated condensations of **11** with aldehyde **5** were unproductive.¹³

Mild transketalization of **14** with bis-(trifluoroacetoxy)iodobenzene¹⁴ gave **15**, which was converted to ketone **16** via oxidative cleavage.¹⁵ Although ketone **16** was susceptible to β-elimination of methanol, Terashima reduction¹⁶ using (+)-*N*-methylphenylephedrine resulted in a remarkably efficient reagent-based hydride addition with high diastereofacial selectivity (8:1 ratio of C₆ alcohols). This new application¹⁷ of the Terashima protocol

(7) Williams, D. R.; Brooks, D. A.; Meyer, K. G.; Pagel, M. *Tetrahedron Lett.* **1998**, *39*, 8023.

(8) Williams, D. R.; Brooks, D. A.; Meyer, K. G.; Clark, M. P. *Tetrahedron Lett.* **1998**, *39*, 7251.

(9) Corey, E. J.; Yu, C.-M.; Kim, S. S. *J. Am. Chem. Soc.* **1989**, *111*, 5495.

(10) Preexisting C₄ asymmetry does not play a significant role in the diastereofacial addition. Model studies of the allylation reaction with achiral Lewis-acids (TiCl₄; CH₂CH₂; –78 °C) produced a mixture of alcohols (40:60 ratio) favoring the corresponding C₈ (*S*)-isomer.

(11) Nishiyama, H.; Yokoyama, H.; Harimatsu, S.; Itoh, K. *Tetrahedron Lett.* **1982**, *23*, 1267.

(12) Synthesis of epoxide **10** proceeds via alkylation of 2-lithio-2-methyl-1,3-dithiane with (*S*)-epichlorohydrin (97% ee; Aldrich) affording net inversion at C₄ (for Mosher ester analysis of **11**; see Supporting Information). Seebach, D. *Synthesis* **1969**, *17*. Braun, M.; Seebach, D. *Chem. Ber.* **1976**, *109*, 669.

(13) Common protecting units, such as esters, thioketals, silyl ethers, and simple ethers are stable to our reaction conditions utilizing bromoborane **13**, whereas acetals and ketals do not survive. Quantitative conversion of **11** to its corresponding bromide required the use of recrystallized *N*-bromosuccinimide with a low temperature, aqueous NaHSO₃ quench to prevent hydrolysis of the 1,3-dithiane.

(14) Stork, G.; Zhao, K. *Tetrahedron Lett.* **1989**, *30*, 287.

(15) (a) Minato, M.; Yamamoto, K.; Tsuji, J. *J. Org. Chem.* **1990**, *55*, 766.

(b) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. *Chem. Rev.* **1994**, *94*, 2483.

(16) (a) Terashima, S.; Tanno, N.; Koga, K. *J. Chem. Soc. Chem. Commun.* **1980**, 1026. (b) Terashima, S.; Tanno, N.; Koga, K. *Chem. Lett.* **1980**, 981.

(17) The use of achiral hydride sources (LiBH₄, NaBH₄, super hydride) or (–)-*N*-methylphenylephedrine provided selectivity slightly favoring the undesired (*S*)-isomer. Interestingly, our Terashima reduction ((+)-*N*-methylphenylephedrine) of the corresponding (*S*)-methyl ether (C₈) of ketone **16** gave predominantly the all-syn arrangement (ratio > 20:1).

(1) (a) Faulker, D. J. *Nat. Prod. Rep.* **1993**, *10*, 497; **1994**, *11*, 395; **1995**, *12*, 135; **1996**, *13*, 435. (b) Kobayashi, J.; Ishibashi, M. *Chem. Rev.* **1993**, *93*, 1753.

(2) (a) Ichiba, T.; Yoshida, W.; Scheuer, P.; Higa, T. *J. Am. Chem. Soc.* **1991**, *113*, 3173. (b) Higa, T.; Tanaka, J.; Kitamura, A.; Koyama, T.; Takahashi, M.; Uchida, T. *Pure Appl. Chem.* **1994**, *66*, 2227.

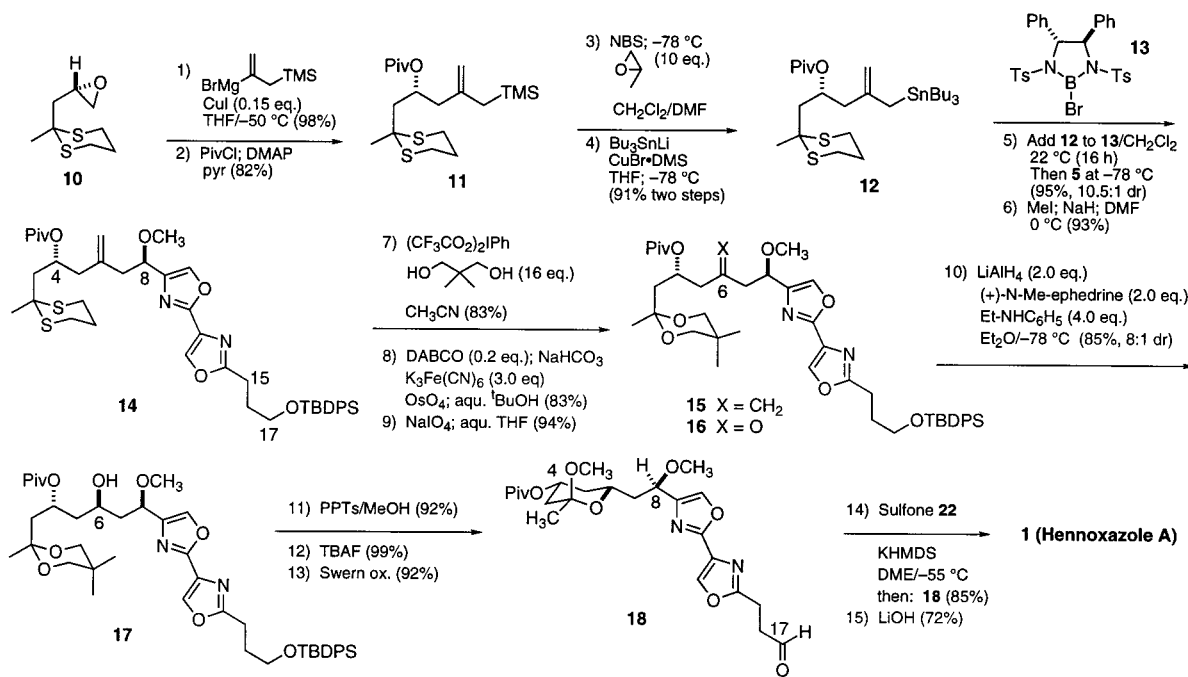
(3) (a) Wipf, P.; Lim, S. *J. Am. Chem. Soc.* **1995**, *117*, 558. (b) Wipf, P.; Lim, S. *Chimia* **1996**, *50*, 157. We thank Professor Wipf for proton and carbon NMR spectra of authentic hennoxazole A for our comparisons.

(4) Binns, F.; Roberts, S. M.; Taylor, A.; Williams, C. F. *J. Chem. Soc., Perkin Trans. I* **1993**, 899.

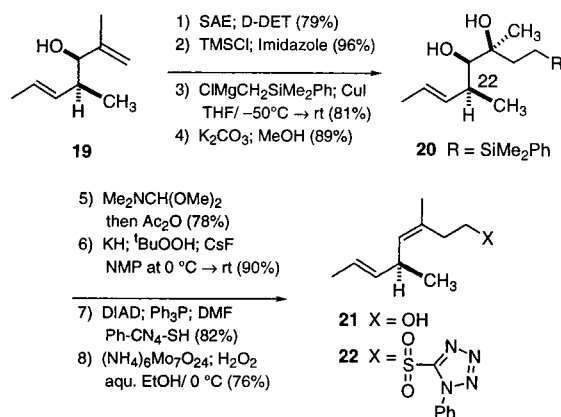
(5) No evidence for the formation of an intermediate fluoride was observed, and fluoride-induced desilylation was not encountered. Lafargue, P.; Guenet, P.; Lellouche, J.-P. *Heterocycles* **1995**, *41*, 947.

(6) Williams, D. R.; Lowder, P. D.; Gu, Y.-G.; Brooks, D. A. *Tetrahedron Lett.* **1997**, *38*, 331.

Scheme 2



Scheme 3



presents opportunities for matched and mismatched reductions of chiral, acyclic β -alkoxyketones. Our studies have important implications for stereocontrolled synthesis of 1,3,5-polyols originating from acetoacetate biogenesis.¹⁸ Finally, the formation of **18** was secured upon mild acidic treatment of **17**, producing a single tetrahydropyran isomer.

The remote C_{22} asymmetry of the nonconjugated triene segment was obtained through chirality transfer in a 2,3-Wittig rearrangement yielding **19** (Scheme 3).^{19,20}

Following a Sharpless asymmetric epoxidation of **19** and protection of the secondary alcohol, nucleophilic oxirane opening

(18) For leading advances in the synthesis of 1,3,5-polyols: (a) Rychnovsky, S. D.; Hoye, R. C. *J. Am. Chem. Soc.* **1994**, *116*, 1753. (b) Rychnovsky, S. D.; Khire, U. R.; Yang, G. *J. Am. Chem. Soc.* **1997**, *119*, 2058.

(19) Abbott Laboratories generously supplied us with quantities of (*S*)-3-buten-2-ol resulting from enzymatic lipase resolution. The alkyne was converted to (2*S*,3*Z*)-3-penten-2-ol by methylation and Lindlar hydrogenation. Alkylation with methallyl chloride (NaH) afforded the substrate for Wittig rearrangement.

(20) Midland, M.-M.; Tsai, D. J.-S. *J. Am. Chem. Soc.* **1985**, *107*, 3915. Review: Mikami, K.; Nakai, T. *Synthesis* **1991**, 594.

(21) Eastwood, F. W.; Harrington, K. J.; Josan, J. S.; Pura, J. L. *Tetrahedron Lett.* **1970**, 5223.

(22) (a) Smitrovich, J. H.; Woerpel, K. A. *J. Org. Chem.* **1996**, *61*, 6044. (b) Molander, G. A.; Nichols, P. J. *J. Org. Chem.* **1996**, *61*, 6040.

(23) Blakemore, P. R.; Cole, W. J.; Kocienski, P. J.; Morley, A. *Synlett* **1998**, 26.

with the Grignard reagent derived from (chloromethyl)dimethylphenylsilane led to 1,2-diol **20**. Syn-elimination to the trisubstituted *Z*-olefin was efficiently carried out via ketalization with *N,N*-dimethylformamide dimethylacetal with subsequent addition of acetic anhydride.²¹ Tamao oxidation of the phenylsilane as described by Woerpel²² provided **21** in 90% yield. Replacement of the primary alcohol with 1-phenyl-1*H*-tetrazole-5-thiol occurred under Mitsunobu conditions. Careful oxidation of the resulting sulfide produced a separable mixture of sulfone **22** (76%) and the corresponding sulfoxides (24%), which were resubmitted to the oxidation conditions to provide additional sulfone (92% overall). Prolonged reaction times resulted in products of olefin epoxidations.

The total synthesis of **1** (Scheme 2) was completed by generation of the α -sulfonyl carbanion of **22** with KHMDS in DME at -55°C followed by addition of aldehyde **18**. Spontaneous elimination of the intermediate β -hydroxysulfone upon warming to ambient temperature facilitates a one-pot generation of C_{17} – C_{18} alkene with excellent *E*-selectivity (*E*:*Z* ratio 91:9) in 85% yield. The Kocienski modification²³ of the Julia–Lythgoe olefination is particularly noteworthy because it provides high trans-stereoselectivity in the absence of factors such as α -chain branching or conjugation. Hydrolysis of the C_4 pivaloate ester (LiOH in aqueous THF/ MeOH (72%)) provided synthetic hennoxazole A (**1**) as a clear, colorless oil; $[\alpha]_D^{25} -46.2^\circ$ (*c* 1.0, CHCl_3), $[\alpha]_D^{25}$ lit. -47° (*c* 3.1, CHCl_3),² identical in all respects with spectra provided for the natural product.³

In conclusion, our convergent route to **1** has offered important advances for asymmetric allylations, efficient oxazole preparations, and a promising new application of the Terashima reduction.

Acknowledgment. Dedicated in memory of Professor George H. Büchi. Financial support was provided by an award sponsored by the National Institutes of Health (GM-41560) and Abbott Laboratories (fellowship to D.A.B.).

Supporting Information Available: Procedures and spectral data for all compounds of the synthesis pathway, and proton NMR spectra for **1**, and compounds **6**–**22** (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.